APPLICATION OF 2D-NMR SPECTROSCOPY IN THE STRUCTURAL DETERMINATION OF GRACILIN B, A BIS-NOR-DITERPENE FROM THE SPONGE SPONGIONELLA GRACILIS.

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Summary. The structure of gracilin B, an unusual metabolite based on a bis-nor-diterpene skeleton isolated from the sponge Spongionella gracilis, has been determined mainly by using two dimensional heteronuclear ^{13}C ^{-1}H shift correlation spectroscopies.

Nor-diterpenes and bis-nor-diterpenes are very rare metabolites from marine origin since only a few examples can be found in the literature $^{2-5}$. In the course of our survey on the constituents of mediterranean invertebrates, we recently isolated gracilin A (1), a nor-diterpene diacetate from the sponge Spongionella gracilis; we wish to report here that this organism also produces gracilin B (2), a highly oxygenated metabolite which, to the best of our knowledge, represents the first bis-nor-diterpene observed from a marine sponge. The new compound (2), $[\alpha]_D$ +191.0 (c = 1.0, CHCl₃), m.p. 167-168°, λ_{max}^{MeOH} = 296 (ϵ = 20150), was obtained in a 2.6% yield from the methanol-chloroform 1:1 extract of the sponge S.gracilis, collected in the Bay of Napoli, by repeated chromatographies over silica gel using dichloromethane and increasing amounts of diethyl ether in $\underline{\textbf{n}}$ -hexane as eluent. Its molecular formula C $_{22}\text{H}_{28}$ O $_{8}$ was deduced by HRMS 7 . The mass spectrum showed a weak molecular ion at m/z 420 (8%) with strong peaks at m/z 360 (M^+ -AcOH, 28%), 318 (M^+ -AcOH-CH₂CO, 100%) and 300 (M^+ -2AcOH, 14%), indicating a facile loss of two acetate groups. The infrared spectrum (CS $_2$) contained bands at 1760,1745 and 1630 cm $^{-1}$. 1 H-NMR spectrum of $\underline{2}$ (250 MHz, CDCl $_{3}$) was particularly detailed and by spin decoupling studies the protons at C-1, C-2 and C-3 [δ 1.34 (2H, bt, J= 5.8 Hz, 3-H $_2$), 1.55 (2H, m, 2-H₂) and 2.19 and 2.40 (1H each, m's, 1-H₂)] and those at C-15, C-10, C-11 and C-16 [δ 3.12 (1H, dd, J= 12.4 and 6.0 Hz, 11-H), 3.95 (1H, dd, J= 12.4 and 6.8 Hz, 10-H), 5.95 (1H, d, J= 6.8 Hz, 15-H) and 6.07 (1H, d, J= 6.0 Hz, 16-H)] were readly inter-related. The protons 7-H and 8-Hwere seen as an AB system at δ 7.17 (J $_{\rm AR} =$ 12.1 Hz) while 12-H and 13-H resonated as two apparent singlets at 6 5.06 and 6.09, respectively. Signals due to the methylene protons 5-H, (6 2.01,2H, s), two acetates (61.99 and 2.02, 3H each, s's) and two Me groups linked to quaternary carbon atoms (& 0.80 and 0.84, 3H each, s's) were also present. Owing to the presence of quaternary

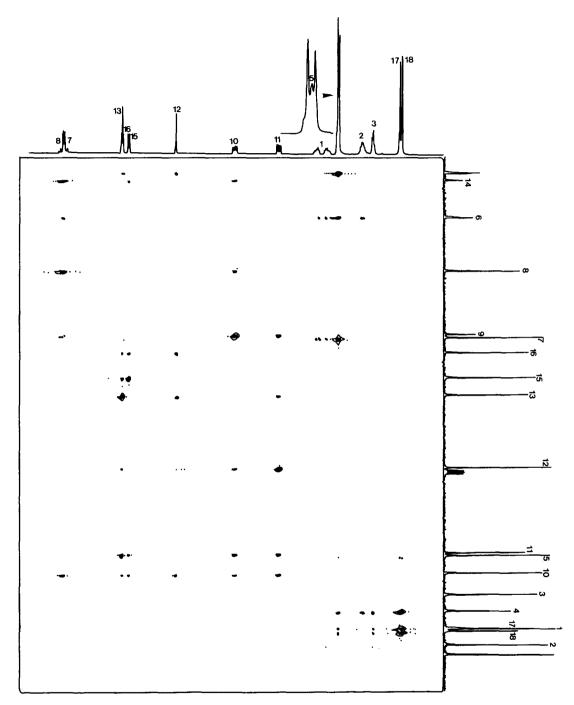


Figure. 62.9 MHz $^{13}\text{C-1}\text{H}$ long range shift correlated 2D-NMR (WM-250) of 2 in CDCl $_3$. The shift correlation with polarization transfer via J-coupling experiment has been carried out with the aid of a Bruker micro-program; fixed delays D $_3$ and D $_4$ were adjusted to give maximum polarization for JCH = 5.0 Hz. In the contour plot some peaks due to J 1 correlation are present, indicating a not complete nulling of direct correlations.

carbon atoms in $\underline{2}$ and to the very small, if any, coupling constants of 12-H with 11-H and 13-H, further structural information could not be obtained by 1 H-NMR spectral analysis. Two dimensional 13 C- 1 H shift correlated spectrum of $\underline{2}$ allowed the assignment of all the signals due to protonated carbons in the 13 C-NMR spectrum of $\underline{2}^{8}$ which also comprised resonances attributable to three \bigcirc C=0 groups and three fully substituted carbons, two of them in the sp 2 region.

Of the numerous structural possibilities which incorporated the above data, structure $\underline{2}$ was confidently assigned based upon 2D-long range ^{13}C ^{-1}H shift correlated spectrum (Figure) which allowed to observe almost all two and three bond ^{13}C ^{-1}H couplings in the molecule. The basic carbon skeleton was established as follows. The only sp 3 quaternary carbon atom (C-4) was seen to correlate with the protons of the two Me groups, with 5-H $_2$, 3-H $_2$ and 2-H $_2$; on the other hand, the carbons resonating at $_6$ 155.7 (C-6) and at 118.7 (C-7) were shown to be long range coupled with 5-H $_2$, 1-H $_2$, 2-H $_2$ and 7-H and with 5-H $_2$ and 1-H $_2$, respectively, thus unequivocally establishing the presence of the substructure $\underline{3}$. The extension of this part structure up to C-14 through C8-C9 was straightforward taking into account UV data and $^1\text{H-NMR}$ spectrum; a confirm came from the presence of a correlation between C-14 and 8-H. This substructure was combined with the above mentioned fragment C15-C10-C11-C16 through the C9-C10 linkage, observing a series of correlation peaks in the 2D-NMR spectrum (Figure), particularly those regarding C-14 with 10-H, C-9 with 10-H and 11-H, C-8 with 10-H and C-10 with 8-H. Ultimately, the positioning of C-12 and C-13 was also made on the basis of long range couplings of C-10 with 12-H, C-12 with 10-H, 11-H and 13-H, C-16 with 12H, and C-13 with 12-H and 11-H.

That the two acetoxy groups were linked to C-12 and C-13 was clearly indicated by long range couplings of the two \supset C=0 groups resonating at & 169.5 and 169.3, with 12-H and 13-H, respectively. What remained to establish to complete structure $\underline{2}$, was the location of the

last three oxygen atoms, which must be comprised by C-14 and C-15, C-15 and C-16 and C-16 and C-13 to form three five-membered rings, since C-14 was seen to correlate with 15-H, C-15 with 16-H, C-16 with 15-H and 13-H, and C-13 with 16-H.

The correct assignment of structure 2 to gracilin B was secured by its reduction with LAH at room temperature for 2h followed by acetylation which gave the expected pentaacetate $\underline{4}$ as the major product; $[\alpha]_D$ +3.5 (c = 0.4, CHCl $_3$), ν $_{max}^{CC1}$ 1750 and 1230 cm , λ $_{max}^{MeOH}$ =249 (ε = 16800), 1 H NMR spectrum (CDCl₃): & 6.34 (1H, d, J= 11.7 Hz, 7-H), 5.90 (1H, d, broadened by long range coupling with 14-H $_2$ and 10-H, J= 11.7 Hz, 8-H), 5.29 (1H, ddd, J= 6.7, 5.0 and 2.6 Hz, 12-H), 4.67 (2H, AB system, broadened by long range coupling with 8-H, $J_{\Delta R}$ =12.4 Hz, 14-H₂), 4.43 (1H, dd, J= 12.4 and 2.6 Hz, 13-Ha), 4.26 (2H, d, J= 6.2 Hz, 15-H₂), 4.10 (2H, AB part of an ABX system, superimposed to 13-Hb signal, $J_{\Delta B}$ =11.7 Hz, 16-H₂), 4.09 (1H, dd, superimposed to 16-H₂ signal, J= 12.4 and 6.7 Hz, 13-Hb), 2.73 (1H, dt, broadened by long range coupling with 8-H, J= 9.9 and 6.2 Hz, 10-H), 2.30 (1H, m, 11-H), 2.16 and 2.14 (1H each, m's, 1-H₂), 2.05 (2H,s,5-H₂), 2.07, 2.04, 2.03, 2.03, and 2.02 (3H each, s's, acetates), 1.54 (2H, m, 2- H_2), 1.34 (2H, t, J= 5.8 Hz, 3-H₂) and 0.87 (6H, s, 17-H₂ and 18-H₂). In the mass spectrum of $\underline{4}$ the parent peak is lacking; diagnostically important peaks at m/z 478 (M⁺-AcOH, 2%), 418 (M⁺-2AcOH, 4%), 376 $(M^{+}-2ACOH-CH_{2}CO, 6\%)$, 358 $(M^{+}-3ACOH, 11\%)$, 316 $(M^{+}-3ACOH-CH_{2}CO, 18\%)$, 298 $(M^{+}-4ACOH, 48\%)$, 256 $(M^{+}-4AcOH-CH_{2}CO, 100\%)$ and 238 $(M^{+}-5AcOH, 100\%)$ are present, indicating the loss of five acetate groups.

The assignment of the overall relative stereochemistry of $\underline{2}$ was accomplished by nOe difference experiments and observing J values in the $^{1}\text{H-NMR}$ spectrum. The Z configuration of the two double bonds was deduced by the positive nOe registered between 7-H and 1-H $_{2}$ and between 8-H and 12-H. The cis relationship among 15-H, 10-H, 11-H and 16-H was also established on the basis of nOe studies. The relative stereochemistry at C-12 and C-13 was based upon an examination of the molecular model taking into account the very small (if any) coupling constants of 12-H with 13-H and 11-H which indicated that the dihedral angles of 12-H with the adjacent protons must be in the range 80-100°.

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- Università di Napoli". 8. 13 C-NMR spectrum (CDCl3): \$169.5 and 169.3 (2 -COCH3), 167.0 (C-14), 155.7 (C-6), 139.2 (C-8), 119.8 (C-9), 118.7 (C-7), 113.9 (C-16), 106.2 (C-15), 100.8 (C-13), 78.5 (C-12), 52.0 (C-11), 51.1 (C-5), 45.8 (C-10), 39.2 (C-3), 34.2 (C-4), 28.5 (C-1), 28.7 and 27.8 (C-17 and C-18), 23.5 (C-2), 20.4 and 20.4 (2CH3CO-).

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